Table 217: $b(E) \times 10^6$ [cm$^2$g$^{-1}$] for Plutonium dioxide (PuO$_2$) $(Z/A) = 0.40583$

<table>
<thead>
<tr>
<th>$E$ [GeV]</th>
<th>$b_{\text{brems}}$</th>
<th>$b_{\text{pair}}$</th>
<th>$b_{\text{nucl}}$</th>
<th>$b_{\text{tot}}$</th>
</tr>
</thead>
<tbody>
<tr>
<td>2.</td>
<td>1.9169</td>
<td>0.2552</td>
<td>0.3693</td>
<td>2.5414</td>
</tr>
<tr>
<td>5.</td>
<td>2.6644</td>
<td>1.7544</td>
<td>0.3937</td>
<td>4.8124</td>
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<tr>
<td>10.</td>
<td>3.2756</td>
<td>2.9226</td>
<td>0.3855</td>
<td>6.5837</td>
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<tr>
<td>20.</td>
<td>3.9032</td>
<td>3.9946</td>
<td>0.3714</td>
<td>8.2691</td>
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<tr>
<td>50.</td>
<td>4.7240</td>
<td>5.6390</td>
<td>0.3557</td>
<td>10.7187</td>
</tr>
<tr>
<td>100.</td>
<td>5.3069</td>
<td>6.7222</td>
<td>0.3483</td>
<td>12.3774</td>
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<tr>
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<td>5.8379</td>
<td>7.6796</td>
<td>0.3448</td>
<td>13.8624</td>
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<tr>
<td>500.</td>
<td>6.4321</td>
<td>8.5554</td>
<td>0.3449</td>
<td>15.3324</td>
</tr>
<tr>
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<td>6.7886</td>
<td>9.0192</td>
<td>0.3503</td>
<td>16.1581</td>
</tr>
<tr>
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<td>9.3523</td>
<td>0.3589</td>
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<td>0.3743</td>
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<td>9.7587</td>
<td>0.3898</td>
<td>17.5951</td>
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<tr>
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<td>7.5299</td>
<td>9.8422</td>
<td>0.4079</td>
<td>17.7800</td>
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<tr>
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<td>9.9042</td>
<td>0.4360</td>
<td>17.9444</td>
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<tr>
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<td>7.6291</td>
<td>9.9300</td>
<td>0.4602</td>
<td>18.0194</td>
</tr>
</tbody>
</table>