Table 246: $b(E) \times 10^6$ [cm$^2$g$^{-1}$] for Silver bromide AgBr

$\langle Z/A \rangle = 0.43670$

<table>
<thead>
<tr>
<th>E [GeV]</th>
<th>$b_{\text{brems}}$</th>
<th>$b_{\text{pair}}$</th>
<th>$b_{\text{nucl}}$</th>
<th>$b_{\text{tot}}$</th>
</tr>
</thead>
<tbody>
<tr>
<td>2.0</td>
<td>1.1873</td>
<td>0.5075</td>
<td>0.3912</td>
<td>2.0861</td>
</tr>
<tr>
<td>5.0</td>
<td>1.6318</td>
<td>1.3843</td>
<td>0.4180</td>
<td>3.4341</td>
</tr>
<tr>
<td>10.0</td>
<td>1.9929</td>
<td>2.0655</td>
<td>0.3993</td>
<td>4.4577</td>
</tr>
<tr>
<td>20.0</td>
<td>2.3636</td>
<td>2.7348</td>
<td>0.3908</td>
<td>5.4892</td>
</tr>
<tr>
<td>50.0</td>
<td>2.8511</td>
<td>3.7278</td>
<td>0.3769</td>
<td>6.9558</td>
</tr>
<tr>
<td>100.0</td>
<td>3.2001</td>
<td>4.3944</td>
<td>0.3687</td>
<td>7.9632</td>
</tr>
<tr>
<td>200.0</td>
<td>3.5216</td>
<td>4.9941</td>
<td>0.3647</td>
<td>8.8805</td>
</tr>
<tr>
<td>500.0</td>
<td>3.8874</td>
<td>5.5528</td>
<td>0.3647</td>
<td>9.8049</td>
</tr>
<tr>
<td>1000.0</td>
<td>4.1110</td>
<td>5.8531</td>
<td>0.3705</td>
<td>10.3347</td>
</tr>
<tr>
<td>2000.0</td>
<td>4.2875</td>
<td>6.0718</td>
<td>0.3797</td>
<td>10.7390</td>
</tr>
<tr>
<td>5000.0</td>
<td>4.4540</td>
<td>6.2559</td>
<td>0.3964</td>
<td>11.1063</td>
</tr>
<tr>
<td>10000.0</td>
<td>4.5377</td>
<td>6.3417</td>
<td>0.4132</td>
<td>11.2925</td>
</tr>
<tr>
<td>20000.0</td>
<td>4.5938</td>
<td>6.3975</td>
<td>0.4327</td>
<td>11.4241</td>
</tr>
<tr>
<td>50000.0</td>
<td>4.6406</td>
<td>6.4398</td>
<td>0.4631</td>
<td>11.5435</td>
</tr>
<tr>
<td>100000.0</td>
<td>4.6618</td>
<td>6.4574</td>
<td>0.4890</td>
<td>11.6083</td>
</tr>
</tbody>
</table>